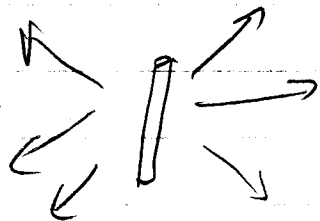


3-12

3rd  
ed



$$P = 10 \text{ W}$$
$$\lambda = 589.3 \text{ nm}$$
$$\Delta t = 1 \text{ sec}$$

$$n = \frac{P}{\frac{hc}{\lambda}} \Delta t = \frac{10 \text{ W} \cdot 589.3 \times 10^{-9} \text{ m} \cdot 1 \text{ sec}}{6.626 \times 10^{-34} \text{ J sec} \cdot 3 \times 10^8 \frac{\text{m}}{\text{sec}}}$$

$$n = 2.96 \times 10^{19} \text{ photons}$$

3-13 3<sup>rd</sup> ed

~~2.12~~

In the photoelectric effect, the kinetic energy,  $K$ , of electrons expelled from a metal surface is

$$K = hf - \phi, \text{ where}$$

$f$  = frequency of the incident photons and  
 $\phi$  = work function of the metal.

A retarding potential or stopping potential is applied which prevents the photoelectrons from escaping. Thus

$$eV_s = K \quad (e = \text{electronic charge})$$

We're given  $\lambda = \frac{c}{f} = 250 \text{ nm}$  and  $V_s = 2.92 \text{ V}$ .

Solve for  $\phi$

$$eV_s = hf - \phi$$

$$\phi = hf - eV_s$$

$$= \frac{hc}{\lambda} - eV_s$$

$$\left[ hc = 1240 \text{ eV nm} \right]$$

$$= \frac{1240 \text{ eV nm}}{250 \text{ nm}} - 2.92 \text{ eV}$$

$$= 2.04 \text{ eV}$$

The work function is  $2.04 \text{ eV}$

$$\left[ 3.27 \times 10^{-19} \text{ J} \right]$$

3-16 3rd ed

~~ZITS~~

## Photoelectric Effect

$$K = hf - \phi = \frac{hc}{\lambda} - \phi$$

Given  $\lambda = 300 \text{ nm}$

$$K = 2.23 \text{ eV}$$

a) solve for the work function

$$\phi = \frac{hc}{\lambda} - K$$

$$= \frac{1240 \text{ eV nm}}{300 \text{ nm}} - 2.23 \text{ eV}$$

$$= 1.90 \text{ eV}$$

For cesium, the work function is 1.90 eV.

b) The stopping potential is given by

$$eV_s = K$$

$$eV_s = \frac{hc}{\lambda} - \phi \quad \text{with now } \lambda = 400 \text{ nm.}$$

$$eV_s = \frac{1240 \text{ eV nm}}{400 \text{ nm}} - 1.90 \text{ eV}$$

$$V_s = 1.2 \text{ Volts}$$

$$3^{-17} \text{ } 3 \sqrt{2} \text{ eV}$$

<del>2-16</del>	$\phi$	eV ( $K_{max}$ )
Li	2.3	1.84
Be	3.9	.24
Hg	4.5	- .71

$$\lambda = 300 \text{ nm}$$

$$eV_s = \frac{hc}{\lambda} - \phi = \frac{4.14 \times 10^{-15} \text{ eV} \cdot \text{m} (3 \times 10^8 \text{ m/sec})}{300 \times 10^9 \text{ m}} - \phi$$

$$eV_s = 4.14 \text{ eV} - \phi$$

a) Li + Be will exhibit the photoelectric effect while Hg will not.

$$b) K_{max} = eV_s \quad \text{m}$$

$$K_{max} = \frac{hc}{\lambda} - \phi = eV_s$$