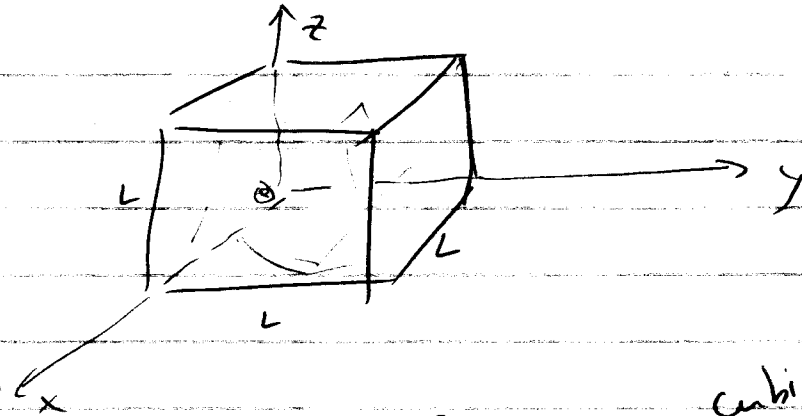


8-2 3K ed
7/2



The electron is confined to a ^{cubical} box of side $L = 0.2 \times 10^{-9} \text{ m}$. The energy levels are

$$E = \frac{\pi^2 \hbar^2}{2m L^2} (n_1^2 + n_2^2 + n_3^2)$$
$$= 9.41 \text{ eV} (n_1^2 + n_2^2 + n_3^2)$$

a) The ground state energy occurs when $n_1 = n_2 = n_3 = 1$, where

$$E_{111} = 3(9.41 \text{ eV}) = 28.2 \text{ eV} = 4.52 \times 10^{-18} \text{ J}$$

b) The first excited state occurs when $n_1 = 2$ or $n_2 = 2$ or $n_3 = 2$

$$E_{211} = E_{121} = E_{112} = 6(9.41 \text{ eV}) = 56.5 \text{ eV}$$

$$= 9.05 \times 10^{-18} \text{ J}$$

3rd ed

8-10

7/10

An electron is in a state

$$n=4, \quad l=3 \quad \text{and} \quad m_l=3.$$

max

$$a) \quad |\bar{L}|^2 = l(l+1) \hbar^2$$

$$= 12 \hbar^2 \Rightarrow$$

$$|\bar{L}| = 3.65 \times 10^{-34} \text{ kg m}^2 \text{ sec}^{-1}$$

$$= 2.28 \times 10^{-15} \text{ eV sec}$$

$$b) \quad L_z = m_l \hbar = 3 \hbar = 3.17 \times 10^{-34} \text{ kg m}^2 \text{ sec}^{-1}$$

$$= 1.97 \times 10^{-15} \text{ eV sec}$$

8-11 3rd ed

7-11 If the ^{orbital} angular momentum of the Earth is quantized (which it is), what is the value of its orbital quantum number, l ?

Given $L = 4.83 \times 10^{31} \text{ kg} \frac{\text{m}^2}{\text{sec}}$, and $L^2 = l(l+1) \hbar^2$.

$$a) \quad l(l+1) \hbar^2 = \left(4.83 \times 10^{31} \text{ kg} \frac{\text{m}^2}{\text{sec}} \right)^2$$

$$l(l+1) = \frac{\left(4.83 \times 10^{31} \text{ kg} \frac{\text{m}^2}{\text{sec}} \right)^2}{\left(1.055 \times 10^{-34} \text{ J} \cdot \text{sec} \right)^2} \quad (\text{units cancel})$$

$$= \left(4.85 \times 10^{65} \right)^2$$

$$l^2 + l = 23.5 \times 10^{130}$$

approximately, $l \approx \sqrt{23.5 \times 10^{130}} \approx 4.85 \times 10^{65}$

$$b) \quad \text{If } l \rightarrow l+1, \text{ the } \Delta L = \left[\sqrt{(l+1)(l+2)} - \sqrt{l(l+1)} \right] \hbar$$

The fractional change is $\frac{\Delta L}{L} = \frac{\Delta L}{\sqrt{l(l+1)} \hbar}$.

$$\text{So, } \frac{\Delta L}{L} = \sqrt{\frac{(l+1)(l+2)}{l(l+1)}} - 1 \approx 0$$

Too small to ever detect.

I mean, too small ever to detect.

Sophisticated persons do not split infinitives.

$$l=2$$

$$R_{2p}(r) = A r e^{-r/2a_0}$$

Expectation value of r $\int_0^{2\pi} \int_0^\pi \int_0^\infty r^4 |A|^2 r^2 e^{-r/a_0} \sin\theta d\theta d\phi dr$

$$\langle r \rangle = 4\pi \int_0^\infty r (A^2 r^2 e^{-\frac{r}{2a_0}}) r^2 dr$$

$$= 4\pi A^2 \int_0^\infty r^5 e^{-\frac{r}{a_0}} dr$$

look up $\int_0^\infty x^n e^{-ax} dx = \frac{n!}{a^{n+1}}$

$$\langle r \rangle = 4\pi A^2 \frac{5!}{\left(\frac{1}{a_0}\right)^6}$$

We need A^2 : $4\pi \int_0^\infty A^2 r^2 e^{-\frac{2r}{2a_0}} r^2 dr = 1$

$$4\pi A^2 \frac{4!}{\left(\frac{1}{a_0}\right)^5} = 1$$

$$A^2 = \frac{1}{4\pi(4!)a_0^5}$$

$$\langle r \rangle = \frac{4\pi 5! a_0^6}{4\pi (4!) a_0^5} = 5 a_0 = 2.65 \times 10^{-10} \text{ m} = 2.65 \text{ \AA}$$

check against Fig 8.11 ...